

# PATENT COOPERATION TREATY

From the  
INTERNATIONAL SEARCHING AUTHORITY

# PCT

To:

see form PCT/ISA/220

## WRITTEN OPINION OF THE INTERNATIONAL SEARCHING AUTHORITY (PCT Rule 43bis.1)

Date of mailing  
(day/month/year) see form PCT/ISA/210 (second sheet)

Applicant's or agent's file reference  
see form PCT/ISA/220

**FOR FURTHER ACTION**  
See paragraph 2 below

International application No.  
PCT/B2004/003968

International filing date (day/month/year)  
02.12.2004

Priority date (day/month/year)  
04.12.2003

International Patent Classification (IPC) or both national classification and IPC  
C09B67/26, C09B67/54, C09B67/22

Applicant  
CLARIANT INTERNATIONAL LTD

### 1. This opinion contains indications relating to the following items:

- ☒ Box No. I Basis of the opinion
- ☐ Box No. II Priority
- ☐ Box No. III Non-establishment of opinion with regard to novelty, inventive step and industrial applicability
- ☐ Box No. IV Lack of unity of invention
- ☒ Box No. V Reasoned statement under Rule 43bis.1(a)(i) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement
- ☒ Box No. VI Certain documents cited
- ☒ Box No. VII Certain defects in the international application
- ☐ Box No. VIII Certain observations on the international application

### 2. FURTHER ACTION

If a demand for international preliminary examination is made, this opinion will usually be considered to be a written opinion of the International Preliminary Examining Authority ("IPEA"). However, this does not apply where the applicant chooses an Authority other than this one to be the IPEA and the chosen IPEA has notified the International Bureau under Rule 66.1bis(b) that written opinions of this International Searching Authority will not be so considered.

If this opinion is, as provided above, considered to be a written opinion of the IPEA, the applicant is invited to submit to the IPEA a written reply together, where appropriate, with amendments, before the expiration of three months from the date of mailing of Form PCT/ISA/220 or before the expiration of 22 months from the priority date, whichever expires later.

For further options, see Form PCT/ISA/220.

### 3. For further details, see notes to Form PCT/ISA/220.

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**Box No. V Reasoned statement under Rule 43bis.1(a)(i) with regard to novelty, inventive step or  
Industrial applicability; citations and explanations supporting such statement**

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1. Statement

Novelty (N)	Yes: Claims	3,4,10-12,14,15
	No: Claims	1,2,5-9,13
Inventive step (IS)	Yes: Claims	10-12
	No: Claims	1-9,13-15
Industrial applicability (IA)	Yes: Claims	1-15
	No: Claims	

2. Citations and explanations

**see separate sheet**

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**Box No. VI Certain documents cited**

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1. Certain published documents (Rules 43bis.1 and 70.10)

and / or

2. Non-written disclosures (Rules 43bis.1 and 70.9)

**see form 210**

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**Box No. VII Certain defects in the international application**

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The following defects in the form or contents of the international application have been noted:

**see separate sheet**

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AUTHORITY (SEPARATE SHEET)**

International application No.

PCT/IB2004/003968

**AP20 Rec'd PCT/PTO 02 JUN 2006**

D2: US -A- 4 955 987  
D3: EP -A- 0 288 434  
D4: EP -A- 0 210 378  
D5: EP -A- 0 114 031  
D6: EP -A- 0 029 960  
D7: DE -A- 3 148 878  
D8: CH -A- 0 648 584  
D9: WO -A- 01/32786

V.1. The present application does not meet the criteria of Article 33(1) PCT, because the subject-matter of claims 1,2,5-9,13. is not new in the sense of Article 33(2) PCT.

V.1.1. D9: WO -A- 01/32786

The direct dyes Direct Blue 75 and Direct Blue 78 are listed in the table on page 13 of D9 as candidates for the ultra filtration process disclosed therein. The steps listed in claim 1 of D9 include an ultrafiltration and a concentration step. No solubilisers, dispersants or tensides are added during the process; storage stable preparations are produced.

Alkanole amines are added in some cases as a base, obviously for the same purpose as in the preferred variation of the current process (page 4, lines 30-33).

It is also a technical problem underlying D9 to prepare storage stable preparations of anionic dyes: "..... Furthermore, on storage, especially at temperatures below room temperature, there are often formed in the concentrated solutions deposits which either cannot be redissolved at all or can be redissolved only by carrying out addition work....." (page 1, par. 1; see also the remarks in the examples).

D9 is rated as a novelty destroying document for claims 1,2,5-9,13.

V.2. Claims 3,4,14,15 are regarded not being inventive.

V.2.1. Direct Blue 71 shows only minor structural differences to the dyes Direct Blue 75 and Direct Blue 78 whereby being used for the same dyeing purposes (e.g. paper dyeing). A skilled person would also apply the process of D9 to D.B. 71 to solve the above mentioned technical problem. Claims 3 and 4 are not inventive over D9.

V.2.2. Claims 14,15 do not contain any features which, in combination with the features of any claim to which they refer, meet the requirements of the PCT in respect of inventive step.

V.3. The subject matter of claims 10-12 seems not to be obvious from the disclosed prior art.

VII. Contrary to the requirements of Rule 5.1(a)(ii) PCT, the relevant background art disclosed in the documents D2-D9 is not mentioned in the description, nor are these documents identified therein.

D2: US -A- 4 955 987

D2 deals with exchanging cations by  $\text{Li}^+$  resp.  $\text{NH}_4^+$  to increase solubility during membrane process and subsequent storage period, thereby avoiding the use of solubilizers, which disadvantages are reported by the D2-authors.

No tris-azo dyes are membrane treated

D3: EP -A- 0 288 434

D3 mentions the optional addition of e.g. solubilizing agents, solvents, surfactants 'and the like' before or during the process for the preparation of concentrated aqueous dye formulations by desalination and concentration by membrane separation. In some examples additives are added, in some not; trisazo dyes are not treated in the membrane process of D3.

D4: EP -A- 0 210 378

D4 (preparing of concn. aq. soln. of anionic dyestuffs by 'cation exchange of soln.' or dispersion with cation exchanger beds to increase solubility) deals with other dyes (trisazos not mentioned) and generally adds solubility enhancers.

D5: EP -A- 0 114 031

D5 [process for the preparation of stable aqueous solutions of water-soluble reactive dyes by membrane separation] also mentions, that, besides buffers, a component which improves the water-solubility of the dye is optionally added to the concentrated and desalted solution (e.g. epsilon-caprolactam, N-methylpyrrolidone etc.)

Trisazo dyes are not mentioned in D5.

D6: EP -A- 0 029 960

D6 deals with concentrated aqueous preparations of reactive dyestuffs (high stability in storage) free of solvents, and a process for their preparation, which comprises a membrane separation process. Urea is proposed as an additive thereby leading to the increased storage stability; the examples mention exclusively monoazo reactive dyes.

D7: DE -A- 3 148 878

Storage-stable, liquid, aqueous dye preparations of reactive dyes are obtained from the as-synthesised dye solutions by a membrane separation process; the D7-process does it obviously without the addition of solubilisers, dispersants resp. tensides. As it seems, buffers are the only additives added to the final solution. D7 is silent about non-reactive trisazo dyes.

D8: CH -A- 0 648 584

A non-reactive trisazo dye is membrane-treated in the process of example 19 of D8. No additional solubilisers, dispersants or tensides are added. The solution is very stable and poor in salt content. The dye differs from those of current claim 1 in that a second SO<sub>3</sub>H group is attached to the right naphthol moiety, two copper cations are linked to give a complex dye and an azoxy group is present in the middle of the molecule. Claims 1 and 9-15 therefore seem to be novel over example 19.

The problem underlying the current application can be seen in 'providing an alternative process for the preparation of storage-stable solutions of anionic trisazo dyes'.

The solution is the process defined in claim 1, which avoids the additional use of solubilisers, dispersants or tensides. The critical technical point of D8 is a process free of organic solvents, whereby the addition of solubilisers is a further option:

"..... Die erfindungsgemässen konzentrierten flüssigen Präparationen können als weitere Zusätze .....Lösungsvermittler wie Caprolactam, Trimethyloläthan, Milchsäureamid oder Tetrahydroxymethylmethan (Pentaerythrit).....[enthalten]...."

Avoiding such chemicals presents the critical technical feature of current claim 1.

D8 therefore seems to lead the skilled person away from current claim 1 by seeking an alternative solution for obtaining storage stable anionic dye solutions.

VII.

VIII. The applicant should deal with the following objections:

VIII.1. On page 8, line 13, a formulation j) is mentioned, which is not listed on the top of the same page [comparative examples run from a) to i)].

VIII.2. There is a contradiction between page 4, lines 30-33 and the comparative examples on pages 7,8 [esp. items i),f)h)]. Current claim 1 clearly excludes solubilisers (as well as dispersants and tensides) as further additives during the process. Alkanolamines resp. their ammonium salts can serve as solubilisers. Though, the addition of such compounds is mentioned on page 4 (as counterion replacement). In the light of the description, page 4, claim 1 therefore is not clear with respect to the addition of

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compounds which could serve as solubiliser.

The applicant should clarify this.

VIII.3. This application deals exclusively with liquid dyestuff preparations; the term liquid should therefore also be introduced into the wording of claim 1 resp. in all independent claims where appropriate.